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Novel antimicrobial finishing of organic cotton fabrics using nano-emulsions derived from Karanja and Gokhru plants



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Abstract

Plant-based antimicrobial finishing of textiles is comparable in efficacy, can replace synthetic antimicrobial agents, and is environmentally safe and effective. However, developing durable antimicrobial finishes on cotton-based textiles is a challenge. This research reports the development and characterization of nano-emulsions obtained from Karanja [Milletia pinnata] and Gokhru [Pedalium murex Linn] plants. The nano-emulsions were produced using Milletia pinnata, coconut oil, and curry leaves (nano-emulsion 1) and Pedalium murex, coconut oil, and curry leaves (nano-emulsion 2). The nano-emulsions were characterized for their thermal stability, particle size, pH, and percentage add-on. Two different oils, with surfactant (polysorbate) ratios [1:1 and 1:2], were finished on organic cotton fabrics using a batch process. Scanning electron microscopy images were evaluated to determine the surface morphology of the finished fabrics, and gas chromatography-mass spectrometry images of nano-emulsions were studied to determine the specific chemical constituents of nano-emulsions I and 2. The finished fabrics were evaluated for their antimicrobial resistance using various gram-positive bacteria [Staphylococcus aureus, Staphylococcus epidermidis], which are found on human skin and cause nosocomial infections, gram-negative bacteria [Escherichia coli and Klebsiella pneumoniae], which cause urinary tract infections, and fungi [Aspergillus niger]. The antimicrobial resistance was in the range of 98.62–99.87%, even after 10 washes, indicating that the finishes were effective and durable. The finished and unfinished fabrics were also evaluated for their performance properties, tensile strength, and moisture vapor transmission rate, and the results indicated good durability and comfort characteristics. Our findings highlight the potential of plant-based antimicrobial agents for durable finishing of cotton textiles with antimicrobial properties, thus preventing the spread of infections.

Keywords

Milletia pinnata, Pedalium murex, nano-emulsions, antimicrobial resistance, antifungal efficacy, wash fastness, tensile strength, moisture vapor transmission rate

The recent global pandemic has raised awareness of hygiene and increased the need to control the transmission of infection through textiles. Textile industry professionals and academics have focused on developing various antimicrobial and antiviral compounds for textiles. Textiles intended for hygiene and medical applications are finished with a range of antimicrobial agents, such as triclosan, metallic salts (TiO₂ and ZnO), chitosan, phenolic compounds (QACs), mainly for the fiber types of cotton, polyester, nylon, wool, acrylic, and polypropylene.¹ Textiles made of natural fibers, such as cotton, wool, and silk, provide excellent

conditions for the growth of bacteria and fungi compared to synthetic filaments.² Bacteria such as *Staphylococcus epidermidis* and *Corynebacterium*

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cause body and clothing odor, whereas Candida albicans and Epidermophyton floccosum cause skin irritation and infections.³ Staphylococcus aureus is a grampositive bacterium that causes community and hospital-acquired infections.⁴ Klebsiella pneumoniae and Escherichia coli are gram-negative bacteria and belong to the Enterobacteriaceae family, which causes community and hospital infections and has high antimicrobial resistance (AMR).⁵ Klebsiella pneumoniae causes pneumonia, sepsis, and inflammation of the urinary tract.⁶ E. coli also cause diseases in the intestinal tract, renal system, and central nervous system.⁵ AMR is a global challenge, and it is estimated that 1.2 million people die from drug-resistant infections.⁷ AMR occurs when bacteria, fungi, and viruses no longer respond to medicines, making them harder to treat, spreading infection and illness, and causing death.⁸

Textiles are finished with antimicrobial agents using several surface modification techniques - continuous and exhaust methods, nanotechnology, microencapsulation, layer-by-layer and vapor deposition, and the sol-gel method.⁹ Antimicrobial agents that generally kill bacteria are called biocidal,¹⁰ and they inhibit the growth of microorganisms, which also prevents odor formation, also referred to as biostatic. In the biostatic mechanism, although it inhibits microorganisms, it may leave the surface with live bacteria, causing transmission of infection. Most antimicrobial agents used in textiles are biocidal.¹¹ The mechanism of antimicrobial activity is based on controlled release and barrier formation mechanisms. In the controlled release mechanism, the antimicrobial agent is incorporated into the textile fibers or chemically bonded, and a gradual release of antimicrobial agent kills the microbes.^{2,12–14} However, the release of the antimicrobial substance decreases in concentration over time (possibly due to repeated laundering), and its effectiveness is lowered. In the barrier formation, the antimicrobial agent bonds onto the textile surface through chemical bonding and they do not leach or release the antimicrobial agent but rather serve as a barrier to the microbes and kill when in contact with the textile surface.13,15

The use of synthetic chemicals (silver-based and mineral-based) as antimicrobial agents may affect health and the environment. For instance, QACs such as benzalkonium chloride are widely used in hand sanitizers and facial cleansers. When released into waterways, these chemicals may not biodegrade easily, and their increased use of QACs may lead to bacteria developing resistance and may spread infections.¹⁶ Many textiles finished with antimicrobials that have intimate contact with the skin affect the skin-derived microbes, causing intolerance of the human body and allergies, and affecting human skin flora.¹⁷ Triclosan, widely

used as an antimicrobial chemical in textiles, is found to be toxic and affects human health (endocrine-disrupting chemical, skin irritations, cancer development).¹⁸ Therefore, it is necessary to identify antimicrobial agents that have the most negligible impact on human health and the environment. The use of plant-based antimicrobial compounds is environmentally friendly, safe, and non-toxic¹⁹⁻²¹ compared with synthetic chemicals and inhibits bacterial pathogens. Several herbal extracts possess bioactive compounds, such as the salvation plant, Salvia officinalis L., which possesses bioactive compounds with antimicrobial, antioxidant, anti-inflammatory, and antiviral effects.²² Bioactive substances in plants include basil (eugenol as the active substance), neem (limonoids such as azadirachtin), and turmeric (curcumin).²³ Active bio components in plants enable an antimicrobial effect on textiles, such as terpenoids, which possess resistance to S. aureus, P. aeruginosa, and V. cholera.^{24,25} Flavonoids possess resistance to bacteria including K. pneumoniae, S. centerica, P. aeruginosa, S. aureus, E. coli,^{26,27} while quinones inhibit S. aureus, B. subtilis, and P. aeruginosa.^{28,29} Plant/herbal extracts from leaves, stems, fruits, roots, seeds, and bark offer antimicrobial properties.³⁰ Previous studies reported antibacterial and antifungal efficacy using neem (Azadirachta indica),³¹ tulsi leaves (Ocimum sanctum),³² aloe vera,³³ and eucalyptus leaves. In this study, two herbal nano-emulsions were prepared using a combinations of herbal oils, namely Milletia pinnata L., curry leaves, and coconut oil, and another combination using *Pedalium murex*, curry leaves, and coconut oil.

Milletia pinnata, commonly called Karanja, belongs to the Fabaceae family,³⁴ which originated in India and is found in parts of Southeast Asia and Australia.³⁵ Karanja is also known as Pongamia glabra, Derris indica, and Pongamia pinnata L.34 Different parts of the Karanja tree (leaves, flowers, fruits, bark, and seeds) are used; they possess medicinal properties and are used for nutritional, medical, and pharmacological applications.³⁶ The leaves of Karanja contain bioactive molecules, such as phenolic acids and flavonoids, which exhibit antimicrobial activity.³⁷ The aqueous methanol extracts obtained from the bark of Karanja exhibited higher AMR, with the zone of inhibition ranging from 16.7 to 26.2 mm, compared with leaves with a range of 10.2-16.2 mm. Authors posit that the antimicrobial activity of aqueous methanol extract of Pongamia pinnata is high in phenolic acids and flavonoids.³⁷ Flavonoids are known to retard the growth of microorganisms through inhibiting nucleic acid synthesis.³⁸ Pongamia pinnata leaf extracts of methanol, chloroform, and ethyl acetate have shown antibacterial effects against food-borne pathogens *B. subtilis, S. aureus, L. monocytogenes, P. aeruginosa,* and *S. typhimurium*. Ethyl acetate and methanol extracts exhibited a strong antibacterial effect against all four gram-positive bacteria, namely *B. subtilis, S. aureus, L. monocytogenes,* and *L. monocytogenes,* with their respective diameter zones of inhibition of 13–18 mm.³⁹ Authors suggested that the antibacterial activity of *Pongamia pinnata* leaf extracts was due to bioactive phytochemicals (flavonoids, alkaloids, flavones, and furanodiketone).³⁹

Pedalium murex, commonly called Gokhru, belongs to the Sesame family and grows in South India, Sri Lanka, Mexico, and Pakistan.⁴⁰ Different parts of the plant - leaves, roots, and seeds - have medicinal value and treat various diseases, such as the urinary tract and reproductive system.⁴⁰ Moreover, it is widely used in Avurvedic medicine to treat various symptoms, including coughs, asthma, skin diseases, and piles.⁴⁰ Earlier studies have shown that aqueous leaf extracts of Pedalium murex showed resistance to fungal growth (Aspergillus niger) with a diameter zone of inhibition of 15 mm.⁴¹ In addition, several studies have reported the phytochemical constituents of *Pedalium murex*, such as fatty acids, flavonoids, triterpenoids, steroids, tannins, saponins, ursolic acid, and vanillin, 42-47 which inhibits microbial growth AMR.

Murraya koenigii (Linn), commonly called curry leaves, is widely used in cuisine due to its aromatic flavor.⁴⁸ Root extracts of *Murraya koenigii* in organic solvents (methanol, chloroform, and hexane) showed good inhibition to microbial growth.⁴⁸ Authors reported that the plant possesses carbazole alkaloids, which are cytotoxic and antioxidative, and have anti-inflammatory properties,^{49–51} and the leaves are rich in mono-terpenoids and sesquiterpenoids, which exhibit antifungal properties.⁵² Curry leaf extracts inhibited the growth of *E. coli* and *S. aureus*, which showed zones of inhibition of 3.2 and 5.9 mm, respectively, when using a 10 mg/mL concentration of curry leaves.⁵³

Coconut oil is an edible oil extracted from the dried kernels of coconuts⁵⁴ with 92% saturated fatty acids. There are different types of coconut oil obtained from different parts of the coconut.55 Coconut oil showed antibacterial activity against Pseudomonas aeruginosa, Escherichia coli, Proteus vulgaris, and Bacillus subtilis.⁵⁶ More recently, virgin coconut oil (VCO) was made from fresh kernels without heating, mainly by fermenting with microbes (Lactobacillus fermentum and Lactobacillus plantarum). VCO contains high lauric acid, 54%, whereas coconut oil has only 2.81%.⁵⁷ Moreover, VCO has lactic acid bacteria (LAB; Lactobacillus plantarum and Lactobacillus paracasei), inhibiting bacterial growth of Pseudomonas Klebsiella aeruginosa, pneumoniae,

Staphylococcus aureus, *S. epidermidis*, and *Escherichia coli*.⁵⁷ It is also interesting to note that coconut oil, when formulated as a cream, is found to exhibit antibacterial and antifungal properties, as the emulsifying agent in the cream aids in the penetration of the active compound monolaurin.⁵⁸

It can be noted from the above review that plantbased constituents derived from Milletia pinnata, Pedalium murex, Murrava koenigii, and coconut oil have the potential to be used as antimicrobial agents. However, none of these herbal extracts have been used in combination and reported for their usage in organic cotton fabrics, particularly in the preparation of nanoemulsion, determining the method of finishing on the fabrics, and evaluating their percentage add-on. In addition, the evaluation of finished fabric performance parameters, including wash durability, antibacterial resistance, antifungal efficacy, tensile strength, and comfort properties, has not been reported before, and it adds to the literature. Therefore, for the first time, this research reports the potential of nano-emulsions derived from these herbal combinations, which produce an eco-friendly antimicrobial finishing of organic fabrics suitable for hygiene applications. Therefore, this research aims to determine the antimicrobial efficacy and performance of GOTS (global organic test standards) certified organic cotton fabrics finished with nano-emulsions produced using mixtures of Milletia pinnata L., curry leaf, and coconut oil, and Pedalium murex, curry leaf, and coconut oil.

Methodology

Materials

Based on the extensive analysis of literature and trials of plant-based herbs and their relative constituents, *Pedalium murex*, *Milletia pinnata*, curry leaves (*Murraya koenigii*), and pure, odorless coconut oil were obtained from Matunga, Mumbai. Polysorbate 80 (glycol), a non-ionic surfactant, was selected for this research and was obtained from Loba Chemicals, Mumbai, and locally sourced organic cotton fabrics, a 20 gsm (grams per square meter) plain-woven fabric and 60 gsm twill-woven fabric, were used for this research and were supplied by Test Fab India (Vapi, Gujarat).

Herbal oil extraction

Fresh leaves of *Pedalium murex* (Gokhru) and *Milletia pinnata* (Karanja) were washed thoroughly with distilled water and dried in an oven at 105°C for 1 h to remove any dirt and impurities. Dried herbs, namely 10 g of *Pedalium murex*, 5 g of curry leaves, and 100 ml of coconut oil, were heated together using steam supplied from a steam generator. The steam distillation process releases the aromatic components of the essential oils.

Solvent extraction

Pure ethanol (99%) was used for solvent extraction; dried herbs were placed in the thimble of a Soxhlet extractor, and ethanol was added. The solvent extract collected was filtered and, as the solvent was evaporated, oil was collected in a pot. The oil yield was calculated using the following formula: [amount of extracted oil (g)/the amount of dry herbs and oil (g)] \times 100%. The above process was repeated for *Milletia pinnata*. Both extracted oils were stored in glass bottles for further analysis.

Preparation of nano-emulsion and characterization

Nano-emulsion was prepared using the herbal oil mixture, which was mixed with the surfactant in the ratios 1:1 (100 ml of distilled water, 1 ml of oil mixture, and 1 ml of polysorbate) and 1:2 and, using a high-speed homogenizer Tool-Tech that mixes at 1000–5000 rpm, the mixture was homogenized for 1h until the nano-emulsion was stable. The particle sizes of nanoemulsion 1 (Milletia pinnata, curry leaves, and coconut oil) and nano-emulsion 2 (Pedalium murex, curry leaves, coconut oil) were assessed using a particle size analyzer (Shimadzu SALD-7500 nano, Kyoto, Japan). The pH of the nano-emulsions was also evaluated using a standard pH meter (EquipTronic) at 37°C. The thermal stability of the nano-emulsions was determined using a Metal-Lab MSI-17B (Metal-lab Scientific Industries, Mumbai, India) at varying temperatures, observing the oil separation from the constituents. The whiteness index of the finished fabrics was evaluated using a SpectraScan 5100+ computer color matching system (Rayscan, Australia), which measures the light reflected from the surface of the fabric at each wavelength to calculate the index.

Gokhru/Karanja extracts was subjected to gas chromatography-mass spectrometry (GC-MS) analysis with an Agilent system, 7890B GC and 5977A MSD detectors with the following parameters: fused silica capillary column $30 \text{ m} \times 0.320 \text{ }\mu\text{m} \times 0.25 \text{ }\mu\text{m}$ thickness. Helium was used as a carrier gas with a 2 ml/min flow rate, average velocity of 52.54 cm cm/s, hold-up time of 0.970 min, and a split ratio of 1:10. The injector temperature was 250°C. The oven temperature was programmed at 160°C with an increase of 5°C/min until 270°C and then 15°C/min to 320°C and a hold time of 5 min at the final temperature of 320°C. Mass spectra were obtained in the range width of m/z 35–600, interface temperature of 290°C, solvent cut time of 3 min, and normal scan speed. In addition, the mass spectra of the separated components were compared with the NIST database (NIST, Version 11), and ChemStation GC/MSD software was used.

Finishing of fabrics with nano-emulsions

Both fabrics were finished with nano-emulsions using the exhaust method (batch process) that uses a Rota dryer machine (Figure 1). The fabrics pass through a winch and are immersed in a bath containing the nanoemulsions. The fabric samples were maintained in exhaustion in a material-to-liquor ratio of 1:50 at 60°C for 1 h. During the exhaust process, the material-to-liquor ratio was trialed at varying temperatures, 60-80°C for 1 h, and it was observed that a more uniform application of nano-emulsion was observed at 60°C for 1h. However, any increase in the temperature caused the nano-emulsions to break down, and the finishing of the fabric was not uniform. This is to allow the penetration of nano-emulsion into the fabric structure. The finished fabric passes through the rollers to squeeze out any excess nano-emulsions, and is washed at room temperature.

Fabric characterization

The surface morphologies of the organic cotton fabrics (20 and 60 gsm) were analyzed using scanning electron microscopy (SEM; Carl Zeiss Supra 40 VP, Oberkochen, Germany); a variable pressure of 30 Pa was used. In addition, energy-dispersive X-ray spectroscopic analysis (EDX) was carried out using an Apollo 40 SDD (Tilburg, Netherlands). EDX enables the identification of the elemental composition of fabric finished with nano-emulsions.

Antibacterial and antifungal resistance tests

Antibacterial tests were evaluated using the quantitative method⁵⁹ (AATCC 100-2019) and the qualitative method (AATCC 147:2016)⁶⁰ with *Staphylococcus aureus* strain no. ATCC 6538 (gram-positive bacteria),



Figure 1. Graphical illustration of the exhaust method.

Escherichia coli strain no. ATCC 10799 (gramnegative bacteria), Staphylococcus epidermidis strain no. ATCC 12228 (gram-positive bacteria), and Klebsiella pneumoniae strain no. ATCC 4352 (gramnegative bacteria). For quantitative tests, 1.0 ml of the test inoculum (S. aureus, E. coli, S. epidermidis, K. pneumoniae) was loaded on fabric swatches of 4.8 ± 0.1 cm diameter. The strains were cultured in nutrient agar and sterilized using an autoclave. After the test samples were allowed to contact for 24 h, the percentage reduction of bacteria was determined using the following formula:

$$R = (B - A)/B \times 100\%$$

where A is the number of bacteria recovered from the inoculated specimen after 24 h and B is the number of bacteria recovered immediately after inoculation, 0 h. A wash test was carried out on the finished samples, and the above quantitative tests were repeated to determine the efficacy after 10 washes (ISO 2 procedure, IS $15370:2005)^{61}$ using a launderometer with standard detergent (IEC), wash temperature of 60° C and material-to-liquid ratio of 1:50. The rinsed and washed samples were dried at room temperature. For qualitative tests, a rectangular specimen (25–50 mm) and control sample are placed in contact with the nutrient agar for 24 h. The zone of inhibition is calculated using the following formula:

$$W = (T - D)/2$$

where *T* is the total diameter of the test specimen and clear zone in mm, *W* is the width of the clear zone of inhibition, and *D* is the diameter of the test specimen. For determining the susceptibility of fabrics to mildew and fungal growth, antifungal tests were carried out using *Aspergillus niger* strain no. ATCC 6275 (AATCC 30: III-2013).⁶² The samples were incubated in a humidity chamber at 28°C and 90% relative humidity for six days. The zone of inhibition is determined around the fabric specimen using a rating scale, where 0 denotes no growth of fungi, 1 denotes traces of growth (10%), 2 denotes light growth (10–30%), 3 denotes medium growth (30–60%), and 4 denotes heavy growth (60–complete coverage).

Fabric performance tests

The finished fabrics were evaluated for their durability to determine their performance post finishing with both nano-emulsions. Tensile strength was determined in the warp and weft directions of woven fabrics, and its breaking extension (%) and elongation (mm) were reported (ASTM D 5035-11 2019).⁶³ The moisture vapor transmission rates (MVTRs) of the finished fabrics were analyzed at a temperature of 32° C, relative humidity of 50%, and air velocity of 0.02–0.3 m/s (ASTM 96/96 M: 2005).⁶⁴

Results and discussion

Particle size, pH, thermal stability, percentage add-on, and whiteness index

As highlighted in the *Preparation of nano-emulsion and characterization* section, nano-emulsions 1 and 2 were prepared following several trials, and in this section, the findings of nano-emulsions 1 and 2 in terms of particle size analysis, pH, thermal stability on nano-emulsions, percentage add-on to finished organic fabrics, and whiteness index of the finished fabrics are reported.

The particle size of nano-emulsions. It can be noted that for both nano-emulsions, the particle size decreased from 'immediately after preparation' to weeks one and two, and this reduction was in the range of 66-70% of the initial particle size in week two. It can be noted for the 1:1 ratio, the particle size for nanoemulsion 1 was in the 106-36 nm region, and for nanoemulsion 2, it was marginally similar in the range of 115-33 nm (Figure 2). The above trend was similar for ratio 1:2 for both nano-emulsions. However, there was a marginal decrease in particle size when the oil-tosurfactant ratio increased from 1:1 to 1:2. Like in previous research,⁶⁵ active antioxidants in herbs reduce the particle size. The addition of the surfactant also reduces the particle size. The surfactant plays a crucial role in the formation of a stable nano-emulsion. It helps in reducing the interfacial surface tension between oil and water by absorbing the liquid-liquid interface (it is the junction between two immiscible liquids). The surfactant has a hydrophilic head and hydrophobic tail. During nanoemulsion formation, the increase in the surfactant results in the decrease of surface tension between oil and water, and it accumulates around the droplets and prevents droplet aggregation throughout the preparation process and storage. Therefore, as the level of surfactant increases, it reduces the surface tension between oil and water and helps form a stable nano-emulsion with the reduction in the particle size.

As the particle size decreases, the ease of penetration into the fabric structure is much better and offers stability and prevents the aerial oxidation of oils that results in malodor and taste.

pH optimization. The optimization of the pH of both nano-emulsions was carried out for both ratios



Figure 2. Particle size analysis.

by adding 0.1% hydrochloric acid and 0.1%sodium hydroxide. The pH for nano-emulsion 1 was 6.07 for both 1:1 and 1:2 ratios, and for nanoemulsion 2, the pH was in the range of 6.2–6.3 for ratios 1:1 and 1:2. Overall, it can be noted that the pH of the nano-emulsions was in the range of 6–6.3, which indicates they are acidic and have a similar pH to that of groundwater.

Thermal stability. The thermal stabilities of nanoemulsions 1 and 2 were studied by placing the beakers containing nano-emulsions 1 and 2 separately in a water bath at a temperature of 95°C. The thermal stabilities of nano-emulsion 1 for ratios 1:1 and 1:2 were 56°C and 57.6°C, respectively, while for nano-emulsion 2, they were 57°C and 58.3°C, respectively. Beyond this temperature range, the nano-emulsions become turbid and break down, and cannot be used again. It should be noted that the nano-emulsions are thermodynamically unstable, and the two immiscible liquids (oil and water) used in this study will break at higher temperatures. In the exhaust process, a uniform application of nano-emulsion was observed at 60°C for 1 h. However, the thermal stability experiments revealed that nano-emulsions 1 and 2 were stable in the region of 56–58.3°C. In the thermal stability experiment, the nano-emulsions were heated directly in a glass beaker, causing them to break down when the temperature was above 58°C; however, in the exhaustion process, the heat was applied with the fabric/nano-emulsion together, and it retained its stability up to 60°C, but any increase in temperature caused the nano-emulsion to break down.

Percentage add-on. In the case of nano-emulsion 1, the percentage add-on was marginally higher for 20 gsm fabric than 60 gsm fabric; this could be related to the fabric structure of the fabrics, the closeness of the weave pattern (fabric count), and the cover factor of the fabric (the area covered by a set of threads) (Figure 3). The SEM images in Figure 4(b) show that yarns are tightly spaced, as indicated in the fabric counts of 60 gsm (117 ends per inch \times 38 picks per inch) and 20 gsm with a more open structure (75 ends per inch \times 27 picks per inch) (Figure 4(a)). The cover factor (K) for 20 gsm was low 21 (warp) + 64 (weft), whilst for 60 gsm, it was marginally high at 63 + 21. The interaction between the fabric surface and nanoemulsions is simple mechanical adsorption without crosslinking. This occurs between the interspaces between fibers/yarns in the fabric structure. This study used two different types of weaves with varying fabric weights to demonstrate percentage add-on for lightweight fabrics. The above trend was similar for nanoemulsion 2. However, when the oil-to-surfactant ratio was increased to 1:2, the percentage add-on was higher for both nano-emulsions 1 and 2. As the surfactant ratio increases, the particle size reduces, which enables the nano-emulsion to disperse within the fabric more easily and uniformly. Overall, it can be inferred that the 20 gsm fabric with the 1:2 ratio had a higher percentage add-on of nano-emulsion than the 60 gsm fabric.

Whiteness index. The whiteness index measurements of finished fabrics with nano-emulsions 1 and 2 revealed marginal variations from unfinished fabrics. The percentage change in the whiteness index for the 20 gsm



Figure 3. Percentage add-on of nano-emulsions on organic fabrics.



Figure 4. Scanning electron microscopy images of organic fabrics: (a) 20 gsm plain weave and (b) 60 gsm twill weave before finishing.

fabric finished with nano-emulsion 1 being marginally higher for the 1:2 ratio compared to the 1:1 ratio. In the case of the 60 gsm fabric, the percentage change in the whiteness index was relatively lower than that of the 20 gsm fabric, with a slightly higher change in the index for the 1:2 ratio than for the 1:1 ratio. As seen earlier in the percentage add-on, the penetration of nano-emulsion was more pronounced for 20 gsm fabric than for 60 gsm fabric due to its open structure and low cover factor, especially for the 1:2 ratio. There was a similar trend for nano-emulsion 2. Overall, it can be noted that the change in whiteness index was well below 10% for both nano-emulsions 1 and 2 and is not noticeable to the naked eye, which is indicated by the dotted blue line in Figure 5.

Antibacterial tests

The quantitative antibacterial tests (AATCC 100) revealed that both nano-emulsions 1 and 2 showed

excellent resistance (the percentage reduction in microorganisms is approximately 99%) to gram-positive and gram-negative bacteria. In the case of the 20 gsm fabric finished with nano-emulsion 1, there is a marginal difference in the percentage reduction of microorganisms for ratios 1:1 and 1:2 for unwashed samples (Table 1). These were in the range of 99.30–99.85%. For the 60 gsm fabric, the percentage reduction was in the range of 99.29–99.89%. In the case of nano-emulsion 2, the percentage reduction of microorganisms was marginally higher than for nano-emulsion 1 in the range of 99.73–99.84% for both 20 and 60 gsm fabrics.

For the 20 gsm fabric, the decrease in the percentage reduction of microorganisms for the 1:1 ratio after 10 washes was in the range of 0.5–0.79%, which is marginal, demonstrating good efficacy in resisting the growth of microorganisms (gram-positive and gram-positive bacteria). In the case of the 1:2 ratio, the drop or reduction in the percentage of microorganisms



Figure 5. Whiteness index: (a) nano-emulsion I (Karanja); (b) nano-emulsion 2 (Gokhru) for 20 and 60 gsm fabric; the blue dotted line indicates the 10% index that is not noticeable to the naked eye (color online only).

after 10 washes was in the range of 0.61–0.78%, which also showed the antimicrobial efficacy after washing. For the 60 gsm fabric, the decrease in the percentage reduction of microorganisms for the 1:1 ratio decrease was in the range of 0.30-0.45% for gram-positive bacteria, whilst for gram-negative bacteria, the decrease was in the range of 0.62-1.04%. A marginal decrease in the percentage reduction of microorganisms (1.04%), especially for gram-negative bacteria, is noted after 10 washes. Generally, the biocidal effect drops slightly. After 10 washes, the antimicrobial agent on the fabric surface washed off from the fabric surface, causing a decrease in antimicrobial reduction. In the case of the 1:2 ratio, for grampositive bacteria, the percentage of reduction was in the range of 0.13-0.42%, and for gram-negative bacteria was in the range of 0.12–0.11%. In the case of nanoemulsion 2, for the 20 gsm fabric, for the 1:1 ratio the decrease in the percentage reduction of microorganisms after 10 washes for gram-positive bacteria was in the range of 0.41-0.67%, and for gram-negative bacteria it was 0.66%. For the 1:2 ratio, the decrease for grampositive bacteria was 0.55–0.82%, and 0.5–0.87% for gram-negative bacteria. For the 60 gsm fabric, with the 1:1 ratio, the decrease in percentage after 10 washes was 0.33–0.54% for gram-positive bacteria and 0.52– 0.91% for gram-negative bacteria; however, in the case of the 1:2 ratio, the decrease was much lower. Overall, it can be inferred that a percentage reduction after 10 washes of less than 1% indicates good wash durability for 20 and 60 gsm fabrics. The Appendix illustrates the antimicrobial efficacy of nano-emulsions 1 and 2, before and after 10 washes. In the case of the parallel streak method, there was no growth of bacteria below the specimen for nano-emulsions 1 and 2, particularly for *S. aureus* and *E. coli*.

Antifungal tests. The antifungal tests were conducted using Aspergillus niger, and as a representative sample, the 1:1 ratio 60 gsm fabric was used for both the nano-emulsions. There was a 4.3 mm zone of inhibition for nano-emulsion 1 and no fungal growth.

		Reduction of r				
		Nano-emulsio	n I (Karanja)	Nano-emulsio	Nano-emulsion 2 (Gokhru)	
Herbal ratio	Test culture	20 gsm	60 gsm	20 gsm	60 gsm	
1:1	S. aureus	99.78	99.74	99.72	99.70	
	E. coli	99.70	99.67	99.71	99.68	
	S. epidermidis	99.85	99.77	99.84	99.71	
	K. pneumoniae	99.71	99.66	99.73	99.69	
1:2	S. aureus	99.80	99.89	99.77	99.85	
	E. coli	99.74	99.85	99.72	99.81	
	S. epidermidis	99.82	99.89	99.84	99.83	
	K. pneumoniae	99.76	99.85	99.74	99.78	
1:1	S. aureus	99.30	99.29	99.31	99.37	
After 10 washes	E. coli	98.96	99.05	99.05	99.16	
	S. epidermidis	99.06	99.47	99.17	99.17	
	K. pneumoniae	98.92	98.62	99.07	98.78	
1:2	S. aureus	99.13	99.76	99.17	99.69	
After 10 washes	E. coli	98.96	99.73	98.90	99.44	
	S. epidermidis	99.21	99.82	99.34	99.74	
	К. pneumoniae	99.08	99.74	98.87	99.24	

Table	ι.	Reduction of microorganisms	(AATCC	100) -	exhaust	method*
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*The standard deviation values were negligible for repeat samples.



Figure 6. Scanning electron microscopy for 20 and 60 gsm fabrics finished with nano-emulsion 1.

However, in the case of nano-emulsion 2, there was no zone of inhibition, and minor traces of fungal growth (rating 1) were noticed.

SEM and EDX

SEM of the finished fabrics with nano-emulsion 1 (Figure 6) reveals that at $1000 \times$ magnification, the round ribbon-shaped cotton fibers were stretched, flattened, and observed to have striations. This can be noticed especially when comparing the Figure 6(a) with Figures 6(c) and (e), and similarly when comparing Figure 6(g) with Figures 6(i) and (k) for the 60 gsm fabric. This visible flattening is more visible for 20 gsm than 60 gsm fabrics, especially for the 1:2 ratio. This stretching of finished fabrics is due to the mechanical adsorption of the fabrics with the nano-emulsions and passing between the nip rollers. The above morphological changes were observed on the fabric/yarn structure for 20 and 60 gsm fabrics finished with nano-emulsion 2 (Figure 7).

Figure 8(a) illustrates the SEM image where a portion of the fabric is examined using EDX to determine the elemental composition of fabric finished with nano-emulsion 1, 1:1 ratio, and for 20 gsm fabric. Three random portions were identified for analysis for each fabric finished with nano-emulsion. Some minor traces of sodium, aluminum, silica, and calcium were observed. This observation was similar for nanoemulsion 2 for the 20 gsm fabric, Figure 8(b), and for the 60 gsm fabric for both nano-emulsions 1 and 2 (Figures 8(c) and (d)). There were also some minor traces of magnesium in the fabric finished with nanoemulsion 2. Generally, most green leaves (spinach) have good magnesium content and are widely consumed. It is worth mentioning that minor traces of silicon and aluminum were also noted. This could have occurred during the finishing process, as the fabric could have picked up when passing between the silicon rubber nip rollers and wet pick aluminum rollers. Based on the analysis, it can be confirmed that there were no harmful elements in the finished fabrics, and they are safe for use.

Tensile strength

In the case of nano-emulsion 1, the tensile strength of the 20 gsm fabrics in the warp direction was reduced for



Figure 7. Scanning electron microscopy for 20 and 60 gsm fabrics finished with nano-emulsion 2.



Figure 8. Energy-dispersive X-ray spectroscopic analysis for 20/60 gsm fabrics finished with nano-emulsions I and 2.

Table	2.	Tensile	strength	(N)
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	Nano-emulsion I Karanja (Milletia pinnata)				Nano-emulsion 2 Gokhru (Pedalium murex)			
	20 gsm fabric		60 gsm fabric		20 gsm fabric		60 gsm fabric	
Sample	Warp	Weft	Warp	Weft	Warp	Weft	Warp	Weft
Control	322.39 ± 11.02	226.01 ± 1.29	$\textbf{662.97} \pm \textbf{2.71}$	$\textbf{287.85} \pm \textbf{12.76}$	$\textbf{322.39} \pm \textbf{11.02}$	226.01 ± 1.29	$\textbf{662.97} \pm \textbf{2.71}$	$\textbf{287.85} \pm \textbf{12.76}$
I:I ratio	$\textbf{170.55} \pm \textbf{33.99}$	$\textbf{231.86} \pm \textbf{20.01}$	$\textbf{626} \pm \textbf{57.93}$	$\textbf{254.21} \pm \textbf{36.03}$	$\textbf{175.90} \pm \textbf{25.80}$	$\textbf{257.86} \pm \textbf{33.14}$	$\textbf{588.29} \pm \textbf{17.23}$	195.91 ± 69.60
1:2 ratio	$\textbf{289.71} \pm \textbf{5.44}$	$\textbf{235.39} \pm \textbf{8.93}$	$\textbf{664.63} \pm \textbf{38.92}$	$\textbf{231.86} \pm \textbf{10.56}$	$\textbf{223.66} \pm \textbf{24.29}$	$\textbf{262.39} \pm \textbf{24.46}$	602.71 ± 0.71	$\textbf{261.92} \pm \textbf{10.36}$
1:1 ratio after 10 washes	$\textbf{289.71} \pm \textbf{5.44}$	$\textbf{235.39} \pm \textbf{18.93}$	$\textbf{616.55} \pm \textbf{23.06}$	$\textbf{265.54} \pm \textbf{9.67}$	$\textbf{278.29} \pm \textbf{4.14}$	187.66 ± 57.93	$\textbf{659.49} \pm \textbf{27.70}$	242.32 ± 13.62
1:2 ratio after 10 washes	317.15±1.60	226.88 ± 2.94	633.40±31.19	$\textbf{259.47} \pm \textbf{10.83}$	$\textbf{234.16} \pm \textbf{3.03}$	319.93±29.55	$\textbf{628.75} \pm \textbf{66.28}$	281.21±17.18

the 1:1 ratio significantly, and for the 1:2 ratio, there was a moderate drop in the tensile strength compared to the unfinished fabric (Table 2). In the case of the 60 gsm fabric (warp direction), there was a gradual reduction in strength for both ratios 1:1 and 1:2. For the 20 gsm fabric in the weft direction, there was a marginal increase in the tensile strength. However, for the 60 gsm fabric, there was a drop in the strength for both ratios. In the case of nano-emulsion 2 and for both 20 and 60 gsm fabrics, the tensile strength was reduced for both ratios for the warp direction. Compared to unfinished fabrics, the tensile strength was reduced post-washing (10 times) and for both ratios in warp and weft directions. Similar to nanoemulsion 1, for the 20 gsm fabric in the weft direction, there was a marginal increase in the tensile strength. It should be noted that the finishing of fabrics with nano-emulsions was due to the simple mechanical adsorption of emulsion on the fabric structure, whereby the nano-emulsions remain between the voids of the

fibers in the fabric. Moreover, the pH of the emulsions was slightly acidic (6.0–6.3), and as cotton fibers are sensitive to acids, a reduction in tensile strength is anticipated.

Moisture vapor transmission rate analysis

The MVTR is defined as the ability of the fabric to allow moisture to penetrate through the fabric over a particular period (ASTM 96/96 M: 2005). The MVTR of finished and unfinished fabrics were evaluated to determine if the finishing of fabrics with nanoemulsions were affected. For both nano-emulsions 1 and 2, 20 and 60 gsm fabrics, for both ratios 1:1 and 1:2, the MVTR was reduced compared to unfinished fabrics (Table 3). This reduction in MVTR also reveals that nano-emulsions were adsorbed by the cotton fabrics, affecting the interspaces between the fibers within the yarn/fabric structure and reducing water vapor transmission through the fabric.

	Gokhru (Pedalium murex	;)	Karanja (Milletia þinnata)		
Sample	20 gsm fabric	60 gsm fabric	20 gsm fabric	60 gsm fabric	
Control I:I ratio	2998.34 ± 18.06 2485.11 ± 124.32	2775.83 ± 76.71 2487.37 ± 102.42	$\begin{array}{c} \textbf{2998.34} \pm \textbf{18.06} \\ \textbf{2458.41} \pm \textbf{3.14} \end{array}$	2775.83 ± 76.71 2113.33 ± 8.0	
I:2 ratio	$\textbf{2472.73} \pm \textbf{83.98}$	$\textbf{2472.72} \pm \textbf{79.19}$	$\textbf{2407.85} \pm \textbf{21.95}$	$\textbf{2465.99} \pm \textbf{2.85}$	

Table 3. Moisture vapor transmission rate $(g/m^2/24 h)$

Table 4. Gas chromatography-mass spectrometry analysis of Karanja (Milletia pinnata)

S. no	Active compounds matched	Retention time (min)	Percentage of total peak area (%)	Notes/comments/chemical formula
I	Octanoic acid	6.863	2.403	Also known as caprylic acid and is a saturated fatty acid and carboxylic acid, CH ₃ (CH ₂) ₆ CO ₂ H. It is an antimicrobial agent. ⁶⁷
2	Dodecanoic acid	11.528	4.077	A saturated fatty acid, also called lauric acid, is found in various plant and animal fats, non-toxic, remains solid at room temperature, melts easily in boiling water, $C_{12}H_{24}O_2$, and has antimicrobial properties. ^{68,69}
3	Tetradecanoic acid	14.096	1.160	Myristic acid – a common saturated fatty acid with for- mula CG3 (CH2) COOH used as an emulsifier and surfactant. ⁷⁰
4	Dodecanoic acid, 2,3-dihydroxypropyl ester	28.755	3.123	C ₁₅ H ₃₀ O ₄ , also called as laurin, 1-mono-; found in coconut oil, effective in killing bacteria. ⁷¹
5	Dodecanoic acid, I,2,3-propanetriyl ester	39.193	2.002	C ₃₉ H ₇₄ O _{6,} also known as trilaurin used in cosmetic products, as occlusive skin conditioning. ⁷²



Figure 9. Gas chromatography-mass spectrometry of the ethanolic extract of Karanja leaves.

GC-MS analysis

GC-MS analysis of Karanja extract revealed (Table 4) the presence of saturated fatty acids – caprylic acid, lauric acid, myristic acid, and 1-mono-laurin. These acids possess antimicrobial properties, which could inhibit the growth of bacteria. In addition, the presence of trilaurin was observed, which is widely used in skin conditioning and cosmetic products, and it reveals that the finishing of nano-emulsion 1 is safe on the skin with its antimicrobial properties. Figure 9 illustrates the GC-MS with peaks (constituents matched against the NIST database), retention time (minutes), and abundance of the constituent in percentage for Karanja extract. Some of the peaks shown in Figure 9 highlight that active compounds matched, and these were octanoic acid (6.91%); dodecanoic acid (11.53%); carbamic acid (28.70%); dodecanoic acid (31.26%); carbamic acid and dodecanoic acid (33.78%). Carbamic acid can be used as a herbicide and insecticide. When used correctly, it can help society prevent insect vector diseases.⁶⁶

Table 5. Gas chromatography-mass spectrometry analysis of Gokhru (Pedalium murex)

S. no	Active compounds matched	Retention time (min)	Percentage of total peak area (%)	Notes/comments
Ι	Dodecanoic acid Tridecanoic acid	6.328	1.064	$C_{12}H_{24}O_2$ – dodecanoic acid is a 12-carbon atom, a saturated fatty acid with bactericidal properties, also found in coconut oil and palm oil. ⁷³ Tridecanoic acid – $C_{13}H_{26}O_2$ – a 13-carbon straight-chain saturated fatty acid. ⁷⁴
2	2-Decenal (E)-	7.723	36.995	C ₁₀ H ₁₈ O – organic compound, which has a role of a nematicide and mutagen. ⁷⁵
3	2-Undecanone	8.133	25.036	CH ₃ (CH ₂) ₈ COCH ₃ – a colorless oil widely found in plant extracts, possesses a strong odor and insect repellent. ⁷⁶
4	Octanoic acid Dodecanoic acid	8.513	11.987	Octanoic acid is also known as caprylic acid and is a saturated fatty acid and carboxylic acid, CH ₃ (CH ₂) ₆ CO ₂ H. It is an antimicrobial agent. ⁷⁷
5	Tetradecanoic acid Undecanoic acid	9.808	1.720	Undecanoic acid is also called undecylic acid, has a toxic effect on fungi. ⁷⁸



Figure 10. Gas chromatography-mass spectrometry of the ethanolic extract of Gokhru leaves.

GC-MS analysis of Gokhru extract (Table 5) revealed saturated fatty acids dodecanoic acid, tridecanoic acid, octanoic acid (caprylic acid), and undecylic acid, which serve as fungicides that can inhibit the growth of fungus. Most of the saturated fatty acids have bactericidal properties. In addition, compounds such as 2-decenal (a nematicide – can be used as pesticide) and 2-undecanone (insect repellent), possess insect repelling properties. Figure 10 illustrates the GC-MS with peaks (constituents matched against the NIST database) and retention times for Gokhru extract. Some of the peaks shown in Figure 10 highlight that active compounds matched, and these were octanoic aid (7.73%); dodecanoic acid (13.11%, 33.3% and 35.62%); and tetradecanoic acid (14.8%). These are fatty acids and dodecanoic acid was observed in higher abundance, which has antimicrobial properties.

Conclusions

Finishing cotton textiles for imparting antimicrobial properties for hygiene applications is one of the top priorities for many textile industries and technical textile professionals. During the current pandemic, public awareness of improving personal hygiene and sanitation was at a peak level. In addition, the increased consumer awareness of the environmental impact of disposable products on the community and the need for durable finishes with environmentally friendly finishing are also of utmost importance.⁷⁹ Antimicrobial finishing with synthetic chemicals could affect health¹⁷ and the environment,¹⁶ and there is a pressing rationale to identify plant-based antimicrobial compounds that can resist a broad spectrum of bacteria and fungi. Plant-based extracts possess bioactive compounds with antibacterial, antifungal, antioxidant, and antiviral properties. In this research, a combination of herbal extracts was identified based on a comprehensive literature survey. Two different nano-emulsions were developed and characterized for particle size, pH, thermal stability, percentage add-on, and whiteness index. Nano-emulsion 1 (Milletia pinnata, curry leaves, and coconut oil) and nano-emulsion 2 (*Pedalium murex*, curry leaves and coconut oil) were mixed with a surfactant to develop two ratios, 1:1 and 1:2. GOTS certified organic cotton fabrics were finished using an exhaust method (batch process).

Results indicated that the particle size decreased from immediately after preparation to weeks 1 and 2. The particle size decreased, enabling it to penetrate the fabric structure easily. The overall change in the whiteness index for both nano-emulsions 1 and 2 and both fabrics was lower than 10%, which is not noticeable to the naked eye. Antibacterial tests for both nanoemulsions 1 and 2 showed excellent resistance to a range of gram-positive and gram-negative bacteria. The percentage reduction of bacteria was 99.30-99.89% for both nano-emulsions 1 and 2 and for both fabrics. The antibacterial efficacy was not affected after 10 washes, showing a <1% reduction in the percentage reduction of microorganisms for both nanoemulsions. This also indicated that the finishing of organic cotton fabrics with herbal nano-emulsions was durable. The 60 gsm finished fabrics with nanoemulsion 1 (1:1 ratio) showed resistance to fungal growth with a clear zone of inhibition. The tensile strength of the finished fabrics (both 20 and 60 gsm fabrics) marginally dropped compared to that of unfinished fabrics, and this drop in the strength values is anticipated with any finishing. SEM analysis revealed minor stretching and flattening of cotton fibers in the case of finished fabrics with no major surface morphological changes. EDX of finished fabrics also indicated there were minor traces of sodium and calcium. The MVTR for both the fabrics and nano-emulsions was reduced after finishing, indicating that interspaces between the fibers in the fabric structure were adsorbed by nano-emulsions.

GC-MS analysis of Karanja and Gokhru extract revealed the presence of saturated fatty acids and bioactive compounds, which have bactericidal and fungicidal properties, highlighting why these finishes on cotton fabrics are antimicrobial. This study adds to the extant literature that demonstrates the potential of the novel combination of herbal nano-emulsions that have durable bacterial resistance to a range of gram-positive and gram-negative bacteria. AMR is a major challenge globally, and if plant-based antimicrobial finishes can inhibit the growth of gram-positive bacteria (S aureus and S epidermidis), gram-negative bacteria (E coli and K pneumoniae) and fungi (A *niger*), which are responsible for the spread of infections through textiles in the community, then this research demonstrates a significant potential of the findings in preventing the transmission of infections in an eco-friendly manner. Furthermore, these finishes using plant-based antimicrobial compounds are easy to finish or replenish, safe next to the skin, and have a negligible impact on the environment compared with synthetic chemicals. Further work on different combinations of herbal constituents and ratios to enhance the antibacterial efficacy of various other blended textiles could open new opportunities for durable and environmentally friendly antimicrobial textiles.

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Appendix: Antibacterial efficacy (AATCC-100) with nano-emulsions 1 and 2



Image I. Antibacterial efficacy (AATCC-100) with nano-emulsion I.



Image 2. Antibacterial efficacy (AATCC-100) with nano-emulsion 2.