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3 **Label-free protein detection based on the heat-transfer method – a case study with the**
4 **peanut allergen Ara h 1 and aptamer-based synthetic receptors**
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28 **Abstract:** Aptamers are an emerging class of molecules which, due to the development of the
29 systematic evolution of ligands by exponential enrichment (SELEX) process, can recognize
30 virtually every target ranging from ions, to proteins, and even whole cells. While there are
31 many techniques capable of detecting template molecules with aptamer-based systems with
32 high specificity and selectivity, they lack the possibility of integrating them into a compact
33 and portable biosensor setup. Therefore, we will present the heat-transfer method (HTM) as
34 an interesting alternative since this offers detection in a fast and low-cost manner and has the
35 possibility of performing experiments with a fully integrated device. This concept has been
36 demonstrated for a variety of applications including DNA mutation analysis and screening of
37 cancer cells. To our knowledge, this is the first report on HTM-based detection of proteins, in
38 this case specifically with aptamer-type receptors. For proof-of-principle purposes,
39 measurements will be performed with the peanut allergen Ara h 1 and results indicate
40 detection limits in the lower nanomolar regime in buffer liquid. As a first proof-of-
41 application, spiked Ara h 1 solutions will be studied in a food matrix of dissolved peanut
42 butter. Reference experiments with the quartz-crystal microbalance will allow for an estimate
43 of the areal density of aptamer molecules on the sensor-chip surface.
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3 **Keywords:** label-free biosensors, biomimetic sensors, heat-transfer method (HTM),
4 aptamers, proteins, Ara h 1.
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1. Introduction

Aptamers are synthetic oligonucleotides which can specifically bind target molecules based on a combination of hydrogen bonding, electrostatic interactions, van der Waals forces and their three dimensional conformation^{1,2}. The selection of the optimal aptamer for a given target molecule is accomplished *in vitro* via a SELEX process (Systematic Evolutions of Ligands by EXponential enrichment) from libraries containing random oligonucleotide sequences^{3,4}. With this procedure, it is possible to develop aptamers for a variety of target categories ranging from ions⁵, to proteins^{6,7}, and even to whole cells^{8,9}. Aptasensors have already been designed based on capillary electrophoresis¹⁰, colorimetric assays¹¹, fluorescence anisotropy¹², surface plasmon resonance¹³, and microgravimetric sensing¹⁴. These techniques ensure a high specificity and selectivity; however they are less suited for integration into portable, low-cost sensor devices. In that aspect, field-effect transducers and electrochemical impedance spectroscopy are interesting alternatives^{15,16,17}. Tran *et al.* described an aptamer-based, impedimetric sensor platform which could specifically detect IgE in buffer solutions and human serum samples¹⁸. These measurements were performed with a high-end impedance analyzer requiring comparatively long measurement times and a refined data analysis based on the phase angle of the impedance signal. More recently, Peeters *et al.* developed an aptamer-based, impedimetric sensor system in pocket format, allowing detecting the peanut allergen Ara h 1 specifically and quantitatively correct in the lower nanomolar regime in buffers¹⁹. The sensor output in this case was simply the amplitude of the impedance signal at a given frequency. While impedimetric measurements allow for a considerable system miniaturization, data acquisition and processing still require a piece of software- and hardware engineering.

Within in this article, we will therefore present an aptasensor based on a purely thermometric concept, the heat-transfer method (HTM). This technique has the benefits of being straightforward and requiring only limited hardware, merely two temperature sensors and a heat-source that can be regulated to keep a pre-defined temperature. This is analogous to a *dc* resistivity measurement in which the electronic current is replaced by a thermal current. Until date, this method has been applied successfully for the analysis of DNA mutations^{20,21}, detection of cells and small organic molecules^{22,23}, and phase transitions in lipids²⁴, but not yet in the context of proteins. The latter is considered as one of the most challenging tasks in label-free biosensing and it is *a priori* unclear whether the recognition of proteins by aptamers brings about measurable changes of the heat-transfer resistance. We

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3 point out that the HTM principle is conceptually different from the approach by Wang *et al.*
4 who detected thrombin with aptamers by determining the binding enthalpy calorimetrically in
5 a sandwich-type assay with detection limits down to 22 nM²⁵. In contrast to this, HTM aims at
6 the detection of molecules from the heat-transfer resistance under steady-state conditions and
7 does not necessitate a temperature ramping of the sample under study.
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11 Until now, the heat-transfer principle is based on empirical grounds and the theoretical
12 foundations are not yet developed. However, there is clearly an interest for heat transport
13 through biological molecules as illustrated *e.g.* by the vibrational-dynamics studies on double-
14 and single-stranded DNA molecules²⁶ and the molecular dynamics simulations of heat flow
15 through lipid membranes²⁷. As a possible starting point for a deeper understanding of the
16 principle behind the HTM technique we note that all aforementioned applications have a
17 common denominator: The enhancement of the heat-transfer resistance goes along with a
18 structural softening at the solid (sensor chip) to liquid interface. In case of DNA this is the
19 transition from 'rigid' double-stranded DNA to highly flexible single-stranded DNA²¹. In case
20 of cell recognition by surface imprints, mechanically flexible membrane material adheres to
21 the polymeric surface²⁸. A jump-like R_{th} increase is observed when lipid vesicles undergo
22 their main phase transition from the gel- to the liquid-disordered phase²⁴. Finally, the small-
23 molecule recognition by molecularly imprinted polymers is based on the weak and reversible
24 binding of these molecules in molecular-size cavities. In summary, these are interface effects
25 in which the molecular layer right at the interface is gaining additional degrees of freedom.
26 Regarding aptamers, we will address below whether the capturing of proteins is also
27 associated with gaining additional degrees of freedom at the solid-to-liquid interface.
28 Throughout all experiments described below, *i.e.* before, during and after capturing of the
29 target proteins, the aptamers are in their natural, temperature- and pH-dependent conformation
30 without utilizing neutralizing oligomers: This concept, introduced recently by Das and
31 coworkers, forces aptamers into a straight and rigid configuration before binding of the
32 molecular targets and seems beneficial for achieving ultralow detection limits when combined
33 with a fluorescence-based readout system²⁹.
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51 For proof-of-principle purposes, we will first show the HTM-based thermal detection of Ara h
52 1 in buffer solutions with an aptamer-based sensor platform. This system was selected for two
53 reasons: First, the molecular recognition between the aptamer and Ara h 1 has already been
54 thoroughly studied³⁰ and, second, it is highly relevant for the food industry as the peanut
55 allergen Ara h 1 (a trimeric protein with 195 kDa molecular weight) is responsible for most
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3 food-related anaphylactic shocks¹⁸. To demonstrate the principle not only in buffer solutions
4 but to provide also a first proof-of-application, experiments will be performed with spiked
5 Ara h 1 samples in a matrix enriched with peanut butter. The sensor platform is generic,
6 making it probably possible to detect a variety of other proteins with only minor
7 modifications, provided that corresponding aptamers do exist. In summary, this is the first
8 sensor platform based on the thermal HTM read-out technique, which can detect proteins in
9 buffers and, moreover, also in more complex matrices. This illustrates that also the molecular
10 recognition of a protein by an aptamer goes along with a measurable heat-transfer effect.
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18 2. Experimental

19 2.1 Materials

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22 The thiol 11-mercapto-undecanoic acid (MUA) and bovine serum albumin (BSA, $M_r \sim 66.5$
23 kDa) were purchased from Sigma Aldrich (Steinheim, Germany) and 1-ethyl-3-(3-dimethyl-
24 aminopropyl)carbodiimide (EDC) from Thermo Scientific (Aalst, Belgium). The peanut
25 allergen Ara h 1 was obtained from INDOOR technologies (Cardiff, Wales) and used as
26 received. The compounds required for the preparation of the buffers, respectively, 2-(N-
27 morpholino)ethanesul-fonic acid (MES buffer, pH 6.0), tris(hydroxyamino)methane-glycine-
28 potassium (TGK buffer, pH 8.3) and phosphate buffered saline solution (PBS buffer, pH 7.4),
29 were purchased from Sigma Aldrich (Steinheim, Germany) and Fisher Scientific (Landsmeer,
30 the Netherlands). Ethanol of analytical grade (anhydrous, purity 99.9%) was from VWR
31 (Leuven, Belgium). The optimal aptamer for Ara h 1 detection had been determined in
32 previous research by Tran *et al.* and the sequence is given below¹³.
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41 Aptamer sequence: $\text{NH}_2 - \text{C}_6 - 5'$

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43 TCGCACATTCCGCTTCTACCGGGGGGGTTCGAGCTGAGTGGATGCGAATCTGTGGG
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45 TGGGCCGTAAGTCCGTGTGTGCGAA 3'

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47 This sequence, consisting of 80 nucleotide bases, was ordered from IDT Technologies
48 (Leuven, Belgium). The 5' end, serving for attachment to the thiols, was modified with an
49 amino group and a C_6 carbon spacer. The structural conformation of the aptamer at selected
50 temperatures, calculated with the online software package 'Mfold'³¹, can be found in ref¹⁸.
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52 As shown previously, this optimized Ara-h1 aptamer shows negligible cross selectivity to
53 proteins with a similar molecular weight such as horse-radish peroxidase (HRP) and BSA¹⁹.
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Even more importantly, reference tests with a second important peanut allergen, Ara h 2 with 17 kDa, have proven negative¹³.

2.2 Aptamer functionalization of sensor chips and protein-sample preparation

The sensor chips, see **Figure 1** for a schematic layout, consisted of gilded silicon substrates ($1 \times 1 \text{ cm}^2$, $450 \mu\text{m}$ thickness) which were prepared as follows: First, a 20 nm adhesive layer of chromium was thermally evaporated under a vacuum pressure of $5 \times 10^{-5} \text{ Pa}$ onto doped silicon chips, followed then by a 80 nm layer of gold. These chips were treated with a Digital PSD series UV-ozone system from Novoscan (Nürnberg, Germany) for 1 h in order to clean the surface and make them more hydrophilic by surface-bound oxygen species. Subsequently, they were briefly exposed to a cold “piranha” solution (H_2O_2 and H_2SO_4 in a 1:3 ratio), rinsed with ethanol, and then incubated for 48 h with a MUA thiol solution in ethanol (concentration 1 mM) at room temperature under nitrogen atmosphere. After cleaning the samples in pure ethanol, the amino-terminated Ara h 1 aptamers were attached via direct EDC coupling in MES buffer of pH 6. This process was monitored *in-real time* by probing the thermal resistance at the solid-liquid interface. In order to reduce non-specific adsorption, especially of proteins, the aptamer-functionalized gold surface was blocked by immersing the substrates overnight into a BSA solution (50 nM in PBS, temperature of $4 \text{ }^\circ\text{C}$), generating a blocking BSA monolayer¹⁸.

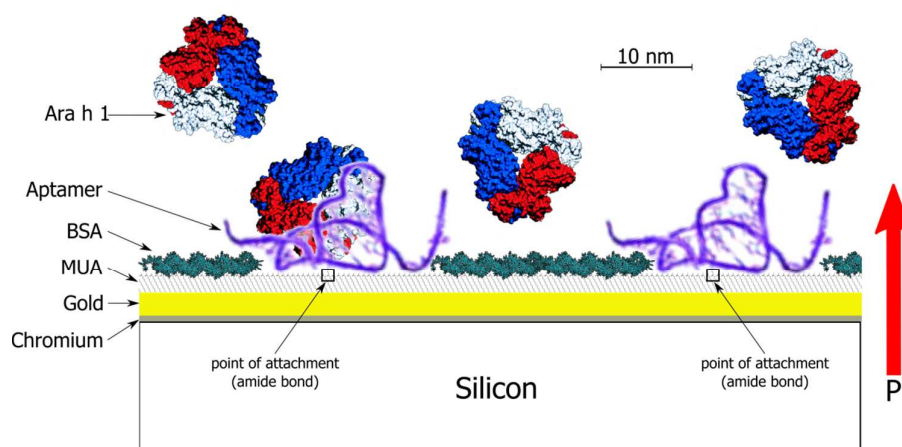


Figure 1: Scheme of the sensor-chip layout in which the amino-modified aptamer is EDC-coupled to a self-assembled monolayer of MUA thiols on gold. The BSA overcoating serves for blocking non-specific adsorption. Under the chosen conditions (TGK buffer with pH 8.3, $37 \text{ }^\circ\text{C}$) the aptamer is expected to attain the indicated conformation before binding of the Ara

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3 h 1 antigen. The flow direction of the heating power P is indicated. The aptamer, BSA, and
4 Ara h 1 are drawn to scale, the scale bar indicates $10 \text{ nm}^{32,33}$.
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8 The aptasensor was now ready for use and, after stabilizing in TGK buffer, various solutions
9 of Ara h 1 concentrations (5, 10, 15, 25, and 50 nM) in TGK buffer were added sequentially
10 to the set up. To address the specificity of the molecular recognition, reference experiments
11 were also performed with a chip with only the SAM and its BSA overcoating, but without
12 presence of the aptamer. As a first proof-of-application, measurements were performed in a
13 matrix enriched with peanut butter. To obtain these samples, 50 mg of peanut butter (Unilever
14 – Calvé, Delft, the Netherlands) was first molten and then dissolved into 200 ml of TGK
15 buffer by stirring for 2 hours at $50 \text{ }^{\circ}\text{C}$. After filtration through a filter with 1 micron pore size,
16 the resulting fluid (also addressed as ‘buffer diluted extract’) was split into two aliquots one of
17 which was unaltered and the other one spiked with 100 nM of Ara h 1. The concentration of
18 Ara h 1 in the non-spiked, buffer-diluted extract can be estimated as follows: The amount of
19 25 mg peanut butter per 100 ml equals 20 mg of pure peanut substance, corresponding to 5
20 mg of proteins. According to literature, the maximum percentage of Ara h 1 in the total
21 protein contents is 16%^{34,35}, meaning that there is maximal 0.8 mg of Ara h 1 present in the
22 100 ml of extract. Together with the molecular mass of Ara h 1 (195 kDa) this corresponds to
23 an upper limit of the Ara h 1 concentration of 40 nM.
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37 *2.3 The thermal resistance set up*

38 The equipment used for the thermal resistance measurements was an in-house design, see
39 Figure 2, which was described previously in detail in ref²⁰. A schematic drawing with the
40 exact dimensions of the equipment can be found in Figure S-1 of the Supporting Information.
41 To this system, a Perspex flow cell with an inner volume of $110 \text{ }\mu\text{l}$ (6 mm diameter, 4 mm
42 and inner height) was coupled, connected to a syringe-driven flow pump. The functionalized
43 sensor chips were mounted horizontally in the set up in order to prevent sedimentation of
44 heavier components, causing possibly non-specific signals by physical adsorption. The sensor
45 chip was hereby pressed mechanically onto the copper block. This block served as thermal
46 reservoir and heat flow was generated using a thermistor on top of the copper block. The
47 temperature of the copper, T_1 , was measured by a thermocouple and actively steered through
48 a PID controller ($P = 8$, $I = 1$, $D = 0$) which in turn was regulating the heating power.
49 Throughout the measurement, this temperature was kept constant at $37.00 \text{ }^{\circ}\text{C}$ in order to
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mimic body temperature and the ambient temperature remained constant at 19.0 °C. The temperature in the liquid, T_2 , was measured with a second miniaturized thermocouple, positioned in the middle of the flow-through cell at a distance of 1.7 mm underneath the chip surface. The thermal resistance (R_{th} , given in °C/W) was then obtained by dividing the temperature difference ($T_1 - T_2$) through the input power P that is required to keep the copper block at the constant temperature $T_1 = 37.0$ °C (Equation I). While the absolute value of R_{th} is governed mainly by the thermal conductivity of the buffer liquid, the positioning of the heated sensor chip at the top of the flow-through cell guarantees that there is no convection of liquid, which could possibly compromise the results.

$$R_{th} = \frac{T_1 - T_2}{P} \quad (I)$$

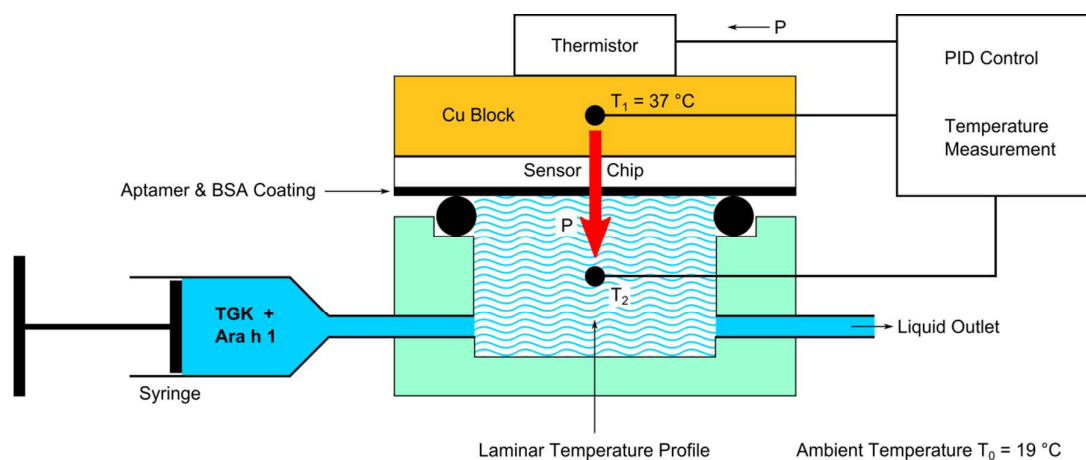


Figure 2: Schematic view of an aptamer-based setup for detecting changes of the thermal resistance upon molecular recognition of proteins. The thermal flow, generated by a thermistor, passes from the backside of the chip through the receptor layer into the sample liquid (heat-flow direction indicated by a red arrow). The temperature gradient is monitored using two thermocouples.

2.4 Microgravimetric measurements

Reference measurements with the quartz-crystal microbalance were performed on a QCM-D E4 system manufactured by Q-SENSE (Gothenburg, Sweden) and data were analyzed with QTools software. We employed UV-ozone treated gold-coated quartz crystals with a nominal

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3 resonance frequency $f_0 = 4.95$ MHz and an active area size of 75 mm^2 . One crystal was left
4 blank, a second was activated with a self-assembled MUA-thiol linker layer and blocked with
5 BSA, and the third chip received the full treatment with the linker layer, tethering of aptamers
6 and finally blocked with BSA as well. All these steps were carried out according to the
7 protocols given in Section 2.2. The measurement temperature (liquid and chip are here at an
8 identical temperature) was $37 \text{ }^\circ\text{C}$ and we evaluated frequency shifts of the fundamental
9 resonance frequency. Due to instrumental constraints, the QCM-sensor chips were mounted
10 horizontally with the active coating facing upward towards the liquid sample.
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18 1. Results and discussion

19 3.1 Thermal monitoring of aptamer functionalization

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22 Prior to the actual protein-recognition experiment, we studied whether the tethering of the
23 aptamers to the thiol-functionalized sensor chips would already bring about a measurable
24 increase of the heat-transfer resistance R_{th} . In previous experiments with DNA sequences
25 from exons of the PAH gene, it was observed that ds-DNA with up to 123 base pairs does not
26 cause an R_{th} increase; this length is slightly below the persistence length of ds-DNA and the
27 fragments are still considered as being ‘mechanically quasi-stiff’. After denaturation to single-
28 stranded fragments consisting of 123 bases, a clear R_{th} jump was observed and explained on
29 grounds of the reduced persistence length of ss-DNA and curling up into Flory spheres^{20,36}.
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37 In the present work, the immobilization of the 80-bases aptamer molecules was
38 performed directly inside the HTM flow-through cell illustrated in Figure 2. First, a baseline
39 was established by mounting a chip, comprising the thiol-linker layer, inside the cell and
40 filling it up with MES buffer. Then, a solution containing MES buffer, aptamer molecules
41 ($0.1 \text{ } \mu\text{M}$), and EDC (400 mM) was added. The temperature T_1 was kept constant at $37.00 \text{ }^\circ\text{C}$
42 throughout the experiment. **Figure 3** presents the measured R_{th} values as bar charts
43 normalized to a baseline defined by the heat-transfer resistance before addition of the
44 aptamers and the EDC reagent. The baseline of 100% corresponds to an absolute value R_{th} of
45 $8.01 \pm 0.14 \text{ }^\circ\text{C/W}$, determined by the materials, dimensions, and interfaces present along the
46 heat-flow path between the two temperature sensors. The height of each bar in Figure 3 was
47 determined as an averaged value over 1000 individual R_{th} data points, taken during a period of
48 1000 s with a sampling rate of 1 point per second. The width of the error bars is given by the
49 standard deviation of the data with respect to the mean value within each of the five
50 considered time intervals. Between 3000 s (50 minutes) and 4000 s (67 minutes) after addition
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of the aptamer-EDC mixture, there is no significant change anymore in R_{th} , indicating that the reaction is completed. This duration of 60 minutes for an EDC-mediated linking reaction of amino-modified nucleotides to COOH groups is similar to the earlier reported optimal duration of approximately 2 hours³⁷. This value of 2 hours was determined with a different monitoring technique (confocal fluorescence microscopy) and on polycrystalline diamond films, a chip material with a higher roughness than the gold layers employed within this study. Therefore, the 60 minutes for optimal EDC coupling found here are in line with the previous results of ref³⁸.

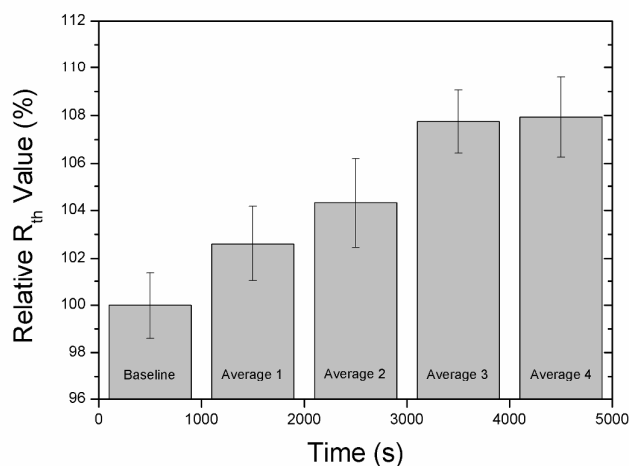


Figure 3: Relative increase of the heat-transfer resistance R_{th} during the EDC coupling of aptamers to the thiol-SAM linker layer. The baseline of 100% (corresponding to an absolute $R_{th} = 8.0 \pm 0.14$ °C/W) was obtained during stabilization in MES buffer. A saturated aptamer coating increases the relative R_{th} to 108 ± 0.8 %. Each bar represents data averaged over a time interval of 1000 s.

The most important observation from Figure 3 is the fact that an aptamer monolayer displays a measurable R_{th} contribution (increase by 0.64 °C/W or 7.9 %). This is similar to single-stranded DNA, even quantitatively regarding the amplitude of the effect, and in contrast to double-stranded DNA where no measurable R_{th} increase occurs even for rather high areal densities of more than 10^{12} ds-fragments per cm^2 ²⁰. Correspondingly, it seems reasonable to assume that aptamers do have a mechanical flexibility irrespective of preferential folding such as indicated in Figure 1.

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3 Next, one may assume that the adsorption of a BSA blocking layer should result in a second
4 increase of the absolute R_{th} value: The adsorption process has to be performed at 4 °C in the
5 refrigerator, meaning that it is not possible to monitor the BSA adsorption in real time
6 because the HTM principle is only applicable when a temperature gradient is present. Instead,
7 we performed a reference experiment with R_{th} monitoring at room temperature while the chip
8 temperature was kept constant at $T_1 = 37.0$ °C. During the adsorption of BSA on a sensor chip
9 covered with only the MUA SAM (no aptamer functionalization) we observed a minor R_{th}
10 increase by 0.1 ± 0.05 °C. This increase stayed persistent even after rinsing and the data are
11 given in the **Supporting Information**, Figure S2. This confirms the strong adhesive
12 properties of BSA while its impact on the heat-transfer resistance is significantly less
13 pronounced as compared to the aptamer.
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23 *3.2 Thermal monitoring of protein recognition and reference testing*

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25 After depositing the BSA overcoating, the sensor chip was installed again in the thermal-
26 resistance setup (Figure 2) and the R_{th} measurement was started after filling the flow-through
27 cell with TGK buffer and allowing stabilizing for 30 minutes. This initial stabilizing
28 guarantees that the system is in thermal equilibrium. Then, Ara-h1 spiked TGK buffer was
29 injected manually with increasing concentrations from 5 to 50 nM. Each spiked TGK sample
30 had a volume of 1 ml, exceeding the inner volume of the flow-through cell 9 times. This way,
31 we ensured that liquid from the previous concentration under study was fully removed from
32 the flow-through cell and the tubing. **Figure 4A** shows a stepwise increase of the heat-transfer
33 resistance for each next-higher concentration while the waiting time before adding the next-
34 higher concentration was chosen 20 to 30 minutes. Immediately after injection of the next
35 concentration there is a temporary overshooting of the R_{th} signal: This is an artifact due to the
36 fact that the injected fluid is at room temperature, taking approximately 5 minutes to achieve
37 again thermal-equilibrium conditions. Therefore, data obtained during the first 5 minutes for a
38 given concentration are discarded. **Figure 4B** displays the concentration dependence of the
39 normalized increase, reaching roughly 9 % for 50 nM of Ara h 1 in TGK. The width of the
40 error bars represents the standard deviation on three separate measurements with the averaged
41 value for a given concentration, ignoring data taken during the first five minutes after
42 injection. For reference purposes, Figure 4B comprises also the normalized R_{th} data for the
43 same Ara h 1 concentrations, but measured with a sensor chip without the aptamer
44 functionality. All other chip features and handling, including the MUA-thiol SAM and the
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BSA overcoating, are identical. Irrespective of the Ara-h 1 exposure, the R_{th} values measured with this reference chip remain constant within the error bars and only for the highest concentration (50 nM) one might infer a slight and non-specific R_{th} increase to $101 \pm 0.8 \%$.

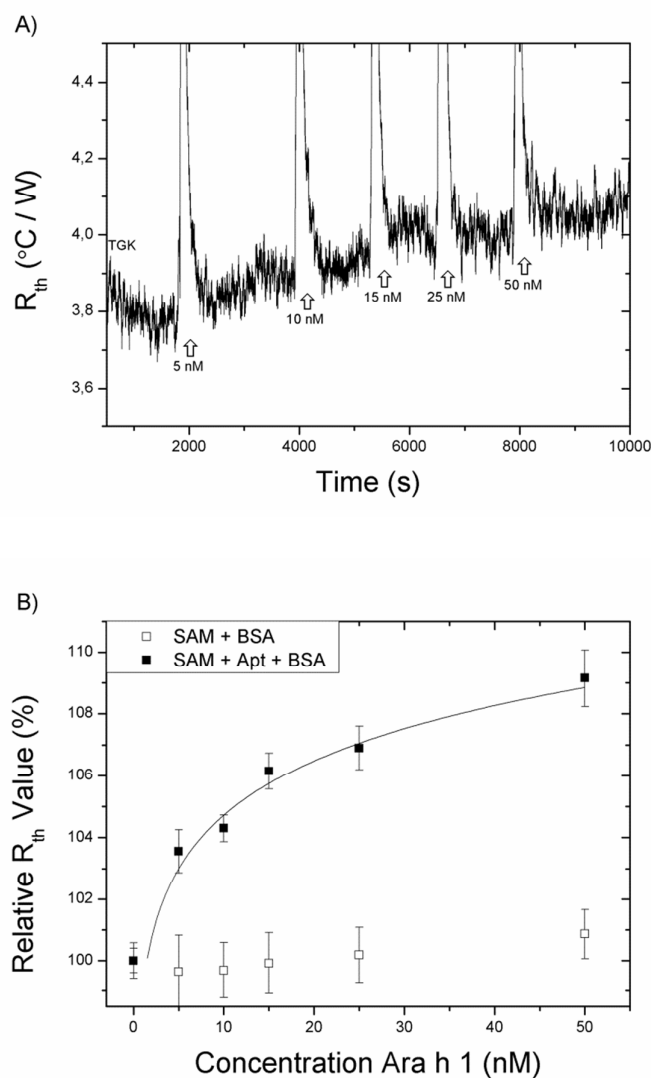


Figure 4: The upper panel (A) shows the absolute values of the heat-transfer resistance of the device with an aptamer-functionalized sensor chip for various concentrations of Ara h 1 in TGK buffer. The displayed data are non-smoothed, raw data. The lower panel (B) presents the data of panel A as solid boxes on a normalized scale with an allometric fit line according to Equation II. Error bars were calculated over three individual measurements with freshly prepared samples. The reference data given as open boxes were obtained with an identically

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3 prepared sensor chip lacking the aptamer functionality. Within error bars, there is no
4 indication for a false-positive response.
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8 At this point it is verified that i) the protein recognition by aptamers enhances the interfacial
9 heat-transfer resistance, ii) that the dose-response curve can give a quantitative information on
10 the protein concentration, and iii) that the BSA overcoating has protein-repellant properties.
11 The observation iii) is evident and expected from daily laboratory practice while i) and ii) are
12 novelties and can thus be utilized as a new technique for label-free protein detection.
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14 From Figure 4B we can now derive an estimate for the limit of detection LOD; the fit curve
15 follows an allometric formula given by Equation II:
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$$y = a \cdot x^b \quad (\text{II})$$

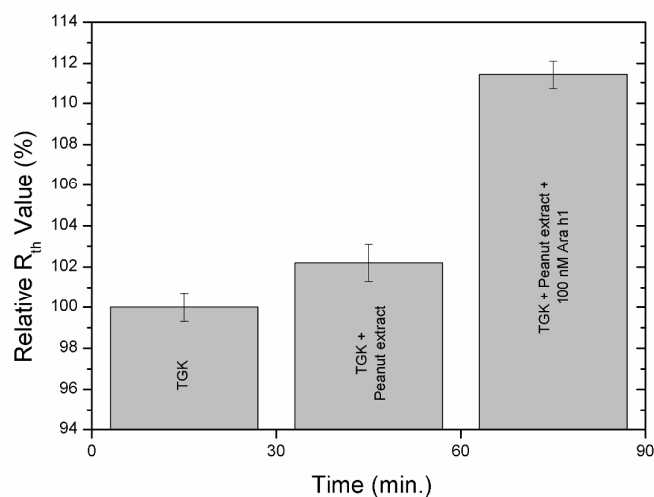
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21 Here, x is the Ara h 1 concentration in nanoMolar units and y is the sensor response in % with
22 100 % corresponding to the baseline. The parameters of the fit curve are $a = 99.1$ %, $b =$
23 0.024 with a coefficient of determination $R^2 = 0.92$. The allometric fit describes the saturation
24 effect occurring at high target molecules concentration and has been used before in ref^{21,23}.
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28 The standard deviation of the signal at the lowest concentration (baseline) is 0.5 %. Defining
29 the LOD as the concentration where the signal amplitude corresponds to the threefold
30 standard deviation we estimate an LOD of 3 nM. The limit of quantification, corresponding to
31 five times the standard deviation, is accordingly found at an Ara h 1 concentration of 4 nM.
32 This agrees very well with the previously obtained value for aptamer-based Ara h 1 detection
33 using impedance spectroscopy as readout technique¹⁸. For comparison, impedimetric protein
34 sensors employing natural antibodies can still reach lower detection limits (< 1 nM), but
35 antibodies are not really competitive with aptamers regarding their price, shelf-life, and
36 reusability³⁹.
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49 50 51 *3.3 Ara h 1 detection in peanut-butter extract*

52 First, a baseline was established in TGK buffer after which the peanut extract (50 mg of
53 peanut butter in 200 ml TGK, see Section 2.2) was added and subsequently the extract that
54 was additionally spiked with 100 nM of Ara h 1. The results are shown in **Figure 5** as bar
55 charts for exposure times 30 min each, discarding data collected during the first 5 minutes
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3 after injection. Upon exposing the functionalized sensor chip to the non-spiked peanut extract,
4 a first increase by $2 \pm 0.8 \%$ was observed. By using the dose-response characteristics in
5 buffer solutions according to Equation 2 this would correspond to an Ara h 1 concentration of
6 $\approx 3 - 5 \text{ nM}$, thus below the maximum of 40 nM estimated from the protein content of the
7 peanut butter. We point out that according to prior literature the Ara h 1 content decreases
8 with increasing roasting time as compared to raw peanuts³⁴. When the spiked solution (100
9 nM of Ara h 1) was introduced, the R_{th} value went further up to $111 \pm 0.7 \%$. Assuming that
10 the analytical form of the dose-response curve (Equation II) stays valid beyond the maximum
11 concentration used for calibration, the increase to 111.3% would correspond to an Ara h 1
12 concentration of 126 nM . This is in good agreement with the value of 100 nM from spiking
13 plus native Ara h 1 present in the peanut butter.
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42 **Figure 5:** Measuring Ara h 1 in a peanut butter enriched matrix: The baseline was obtained in
43 pure TGK buffer and the sample obtained from dissolved and filtrated peanut butter displays a
44 measurable R_{th} increase by $\pm 2.2 \%$. The sample spiked with 100 nM of Ara h 1 (the total
45 Ara-h1 contents slightly exceeds 100 nM) shows a substantial R_{th} increase by almost 12% as
46 compared to baseline level.
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53 This is a first proof-of-application showing that screening of allergens can also be performed
54 in liquefied and filtrated food samples. Furthermore, the magnitude of the R_{th} increase is
55 surprisingly high comparing the size of the molecules that are bound to the solid-to-liquid
56 interface to the overall macroscopic dimensions of the heat-flow path.
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3.4 Comparative study with quartz-crystal microbalance QCM

Since the HTM-based results for protein detection are the first of their kind, additional control experiments were performed with the QCM, a well-established bioanalytical technique. For these experiments, we employed three Au-coated, standard QCM crystals: The first crystal was used as delivered with a non-modified gold surface, the second crystal was functionalized with a SAM-layer of MUA thiols and subsequently treated with a BSA blocking layer. The third crystal was functionalized with the SAM, followed by EDC linking of the aptamers, and finally overcoated with a BSA layer. All surface modifications were carried out according to the protocols given in Section 2.2 for the preparation of the HTM sensor chips. These references were used in order to verify the efficacy of the sensor coating independently and also to check for matrix-related viscosity effects due to the presence of peanut butter. Furthermore, there is a relevant difference with HTM because the functionalized quartz-crystals have to be mounted horizontally in the QCM device with the functionalized surface pointing upward. Therefore, the QCM data can in principle be affected by sedimentation of microparticles or heavier molecules. The experiment was conducted according to the same procedure as for the HTM experiments described in Section 3.3: First, the signal was stabilized in TGK buffer; thereafter, the non-spiked peanut extract was added, and finally the three different QCM chips were exposed to the peanut extract spiked with 100 nM of Ara h 1. The QCM-response of the three differently prepared quartz crystals to these three different solutions is summarized in **Figure 6**.

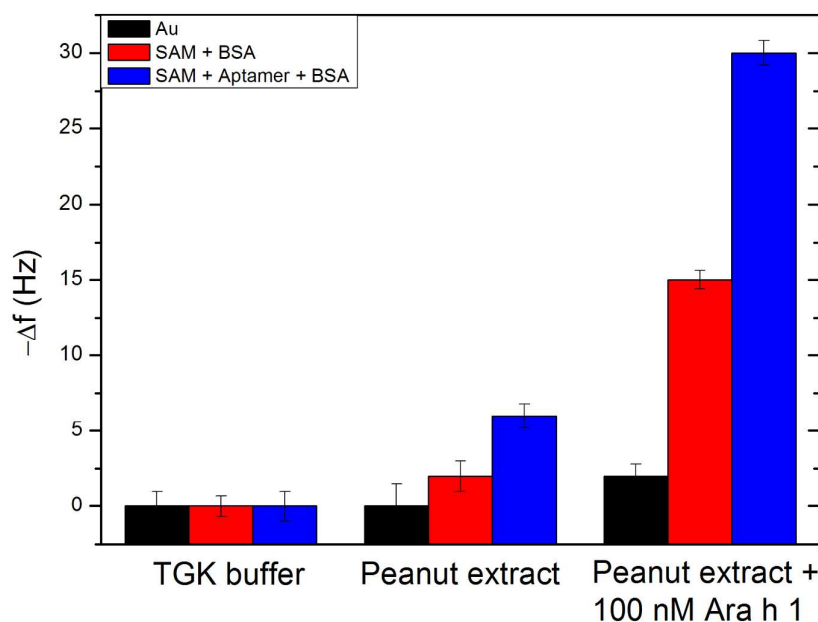


Figure 6: Shift of the resonance frequency Δf of the quartz crystals up exposure to pure TGK buffer (baseline), to the non-spike peanut extract, and to peanut extract spiked with 100 nM Ara h 1. The black bars correspond to non-treated quartz crystals with gold coating, the red bar represent crystals coated with the MUA, SAM and BSA, and the blue bars represent samples with MUA, SAM, aptamer functionalization and the BSA blocking layer.

The addition of the peanut extract or the extract spiked with Ara h 1 did not result in a significant frequency shift when the Au-coated QCM crystal was left non-functionalized. For the spiked sample we see a shift of 2 ± 1 Hz which is not there in case of the non-spiked extract. From this we can at least conclude that the measurements are not affected by the slightly enhanced viscosity of the extract as compared to pure TGK buffer. In case that the QCM crystals were treated with a SAM layer blocked with BSA, the crystal became more hydrophilic and was more prone to non-specific absorption. With only the peanut extract the increase was insignificant, but when spiked with 100 nM Ara h 1, a decrease by 15 ± 1 Hz was measured. This was however still significantly lower compared to the fully aptamer-functionalized crystal: In that case with the extract already a significant decrease by 6 ± 1 Hz was observed and with the spiked extract this resulted in a frequency drop by 30 ± 1 Hz. By taking the difference between the reference experiment (crystal with SAM and BSA, only

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3 non-specific adsorption) and the sample functionalized with aptamer (non-specific adsorption
4 and specific recognition together), the signal attributed to specific binding of Ara h 1
5 corresponds to 15 Hz. Here, we refer to the data obtained with the spiked extract and we not
6 that the observed frequency shifts remained stable even after rinsing with TGK buffer.
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11 If we assume that the Sauerbrey equation is valid, this frequency response equals a mass
12 addition of roughly 204 ng or $6.2 \cdot 10^{11}$ specifically bound Ara h 1 molecules. Due to the high
13 spiked Ara h 1 concentration we can furthermore assume that each of the tethered aptamer
14 molecules captures one Ara h 1 protein. In combination with the active surface area of the
15 QCM-sensor chips of 75 mm^2 we can deduce an areal density of approximately $8 \cdot 10^{11}$ bound
16 aptamers per square centimeter. This is in the same order as the earlier determined surface
17 density of double-stranded DNA fragments tethered to synthetic diamond layers with the
18 fatty-acid & EDC coupling technique, being widely analogous with the protocol described in
19 Section 2.2^{20,38}. Furthermore, we can give a rough estimate for the Flory radius r_F of an
20 aptamer based on Equation III when ignoring for simplicity its self-folding properties²⁰:
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$$r_F \approx \sqrt{\frac{l_p \cdot L}{3}} \quad \text{(III)}$$

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32 The persistence length of single-stranded DNA is $l_p = 1.5 \text{ nm}$ and the fragment length is $L =$
33 28 nm for 85 bases. This results in a Flory radius $r_F \approx 3.7 \text{ nm}$ or $2.3 \cdot 10^{12}$ aptamers per cm^2 ,
34 being approximately three times more than our indirect evaluation based on the QCM results.
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39 The data illustrate that also QCM allows detecting Ara h 1 in a spiked but complex matrix.
40 The QCM responds also to the non-spiked peanut extract; however the signal does not exceed
41 the LOD defined by the threefold noise level and is partially due to non-specific adsorption
42 effects. The adsorption effect is noticeable in the spiked- as well as in the non-spiked peanut
43 extract which can be attributed to the horizontal configuration with the QCM chip underneath
44 the fluid under study.
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52 **4. Conclusions**

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54 The heat-transfer method (HTM) monitors the properties at the solid-liquid interface and has
55 been used in the past for the detection of cells, small organic molecules, and phase transition
56 in lipids. We have shown for the first time the possibility to determine also protein
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3 concentrations with this technique. This was demonstrated by employing an aptamer-based
4 receptor system for the peanut protein Ara h 1. First, the functionalization procedure was
5 followed *in situ*, confirming the presence of the aptamer on the gold surface, which had been
6 pre-activated with a MUA-thiol linker layer. Then, proof-of-principle measurements were
7 performed with Ara h 1 in buffer solutions. A dose-response curve was constructed up to Ara
8 h 1 concentrations of 50 nM and the detection limit was in the order of ~ 3 nM, comparable to
9 optimized values obtained by impedance spectroscopy. As a first proof-of-application, peanut
10 extract and peanut extract spiked with Ara h 1 were studied, demonstrating the possibility to
11 screen for trace allergens in liquefied and filtered food matrices. A distinguishable signal was
12 already obtained with non-spiked peanut extract in which 50 mg of peanut butter was
13 dissolved in a 20.000 times higher volume of TGK buffer.
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23 To verify these results independently, additional experiments were conducted with the quartz-
24 crystal microbalance QCM: Also this microgravimetric technique is capable of detecting Ara
25 h 1 in a food matrix using aptamer receptors; however, it is considerably harder to integrate
26 QCM in a miniaturized, portable and cost-efficient device. Nevertheless, the QCM
27 experiments provided a very reasonable and realistic estimate for the areal density of aptamer
28 receptors on gold surfaces in the order of $8 \cdot 10^{11}$ molecules per square centimeter. We point
29 out that HTM requires especially little instrumental equipment while it can detect proteins
30 specifically and label-free even in complex matrices. The combination of a blocking layer
31 with an upside-down positioning of the sensor chip allowed to suppress non-specific
32 adsorption to the widest possible extend. Moreover, the aptamers stayed at any time in their
33 intrinsic conformation without necessitating the use of neutralizing oligomers, which are
34 displaced upon binding of the target proteins. We assume that the described methodology
35 offers a new approach for the quantitative detection of proteins including allergens and
36 applications can be seen in biomedical- and clinical research as well as in food-safety
37 screening.
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49 **Supporting Information Available**

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52 In the Supporting Information, a design with the exact dimensions of the heat-transfer
53 resistance set up is provided and the effect on the thermal resistance when a BSA overcoating
54 is applied onto the sample in order to minimize non-specific binding. Supporting information
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3 to this document contains additional graphs that are referred to in the text. This material is
4 available free of charge via the Internet at <http://pubs.acs.org>.
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20 **Author Contributions**

21
22 M.P. developed the recognition layer, prepared all samples and wrote the manuscript. The
23 heat-transfer device was designed by B.v.G, and all heat-transfer measurements were performed by
24 B.v.G. in cooperation with M.P. and T.C. B.v.G. and M.P. interpreted the heat-transfer data in close
25 cooperation with W.D.C., R.T. and P.W.. K.L.J.M, P.C. and J.L assisted with the modeling of the
26 aptamers and Ara h 1. E.R.P. provided the aptamers and assisted with questions regarding the
27 immobilization protocols and conformational aspects. G.W. performed the heat-transfer measurements
28 of the BSA overcoating. The manuscript was written by M.P. in collaboration with P.W. All authors
29 have given approval to the final version of the manuscript.
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Figure Table of Content:

